

### **NEET RESULT 2023**



**JAY PITHADIYA** 

639/720



**SUJAL PARMAR** 

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**MINAL PATIL** 

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**ROSNA THOMAS** 

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RITU PATEL

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**RIA CHANDARANA** 

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**KACHHATIYA DHAVAL** 

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**RUTVI PAREKH** 

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**SHREJA GUPTA** 

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37 OUT OF 53 STUDENTS SCORED MORE THAN 450+ MARKS.

### IMPORTANT INSTRUCTIONS

#### **GENERAL INSTRUCTIONS**

- 1. This booklet is your Question Paper.
- 2. Blank papers, clip boards, log tables, slide rule, calculators, mobile or any other electronic gadgets in any form are not allowed to be used.
- 3. Write your Name in the space provided in the first page of this booklet.
- **4.** No rough sheets will be provided by the invigilators. All the rough work is to be done in the blank space provided in the question paper.
- 5. No query related to question paper of any type is to be put to the invigilator.

#### **INSTRUCTIONS FOR OPTICAL RESPONSE SHEET (ORS)**

- > Darken the appropriate bubbles on the original by applying sufficient pressure.
- > The original is machine-gradable and will be collected by the invigilator at the end of the examination.
- Do not tamper with or mutilate the ORS.
- Before answering the paper, fill up the required details in the blank space provided in the Objective Response Sheet (ORS).
- Use a BLACK / BLUE BALL POINT to darken the bubbles in the ORS sheet.
- Darken the bubble COMPLETELY.
- > Darken the bubble **ONLY** if you are sure of the answer.
- The correct way of darkening a bubble is as shown here :
- ➤ There is **NO** way to erase or "un-darkened bubble.
- The marking scheme given at the beginning of each section gives details of how darkened and not darkened bubbles are evaluated.

Marks distribution of questions is as follows.

		Vibro'NET				
	Subject		Marks to be awarded			
S.No.		Nature of Questions	No. of Questions	Correct	Wrong	Total
1 to 25	PART-I (Physics)	Single Choice Questions (SCQ)	25	4	0	100
25 to 50	PART-II (Chemistry)	Single Choice Questions (SCQ)	25	4	0	100
50 to 100	PART-III (Biology)	Single Choice Questions (SCQ)	50	4	0	200
		Total	100			400

Zero marks '0' If none of the options is chosen (i.e. the question is unanswered).

|--|

A ball is dropped into a well in which the water level is at a depth h below the top. Let the speed of sound be c. then the time after which the splash is heard will be given by



- (B)  $h \left[ \sqrt{\frac{2}{gh}} \frac{1}{c} \right]$
- (C)  $h\left[\frac{2}{g} + \frac{1}{c}\right]$
- (D)  $h\left[\frac{2}{g} \frac{1}{c}\right]$

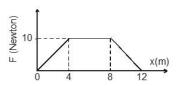
The three initial and final position of a man on the x-axis are given as :

- (i) (-8m, 7m)
- (ii) (7m, -3m) and
- (iii) (-7m, 3m)

Which pair gives the negative displacement?

- (i)
- (B)
- (C) (iii)
- (I) and (ii)
- A body in equilibrium will not have:
  - (A) Velocity
  - (B) Momentum
  - (C) Acceleration
  - (D) All of the above
- 4. In a tug of war each of the two teams apply 1000 Newton force at the ends of a rope, which is found to be in equilibrium, the tension in the rope is
  - (A) 2000 newton
  - (B) 1000 newton
  - (C) 500 newton

5. A particle of mass 0.1 kg is subjected to a force which varies with distance as shown in figure. If it starts its journey from rest at x = 0, its velocity at x = 12 m is

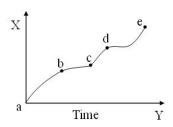


(A) 0 m/s

(D) zero

- (B)  $20\sqrt{2} \text{ m/s}$
- (C)  $20\sqrt{3}$  m/s
- (D) 40 m/s
- 6. The displacement is given by  $x = 2t^2 + t + 5$ , the acceleration at t = 5 sec will be
  - (A)  $8 \text{ m/s}^2$
  - (B)  $12 \text{ m/s}^2$
  - (C)  $15 \text{ m/s}^2$
  - (D)  $4 \text{ m/s}^2$
- A body falls from a height h = 200m. The ratio of distance travelled in each 2 sec during t= 0 to t= 6 second of the journey is
  - (A) 1:4:9
  - (B) 1:2:4
  - (C) 1:3:5
  - (D) 1:2:3
- Two balls are dropped from heights h and 2h respectively from the earth surface. The ratio of time of these balls to reach the earth is
  - (A)  $1:\sqrt{2}$
  - (B)  $\sqrt{2}:1$
  - (C) 2:1
  - (D) 1:4
- 9. A body falls from rest in the gravitational field of the earth. The distance travelled in the fifth second of its motion is  $(g = 10m/s^2)$ 
  - (A) 25 m

- (B) 45 m
- (C) 90 m
- (D) 125 m
- 10. The displacement versus time graph for a body moving in a straight line is shown in figure. Which of the following regions represents the motion when no force is acting on the body?



- (A) ab
- (B) bc
- (C) cd
- (D) de
- 11. The initial velocity of a particle is u (at t = 0) and the acceleration f is given by at. Which of the following relation is valid?
  - $v = u + at^2$
  - (B)  $v = u + a \frac{t^2}{2}$
  - v = u + a
  - v = u
- 12. For a moving body at any instant of time
  - (A) If the body is not moving, the acceleration is necessarily
  - (B) If the body is slowing, the retardation is negative
  - If the body is slowing, the distance is negative
  - If displacement, velocity and acceleration at that instant
    (D) are known, we can find the displacement at any given time in future.

 A particle moves along x-axis in such a way that its coordinate X varies with time t according to the equation

 $x = \{2 - 5t + 6t^2\}$  m. The initial velocity of the particle is

- -5 m/s
- (B) 6 m/s
- -3 m/s
- (D) 3 m/s
- 14. The work done against gravity in taking 10 kg mass at 1 m height in 1 sec will be
  - (A) 49 J
  - (B) 98 J
  - (C) 196 J
  - (D) None of these
- 5. If the kinetic energy of a body increases by 0.1%, the percent increase of its momentum will be
  - (A) 0.05%
  - (B) 0.1%
  - (C) 1.0%
  - (D) 10%
- 16. If the momentum of a body is increased n times, its kinetic energy increases
  - (A) n times
  - (B) 2n times
  - (C)  $\sqrt{n}$  times
  - (D) n<sup>2</sup> times
- 17. An object of mass 3m splits into three equal fragments. Two fragments have velocities  $\hat{\vec{y_j}}$  and  $\hat{v_i}$ . The velocity of the third fragment is
  - (A)  $v(\hat{j} \hat{i})$
  - (B)  $v(\hat{i} \hat{j})$
  - (C)  $-\mathbf{v}(\hat{\mathbf{i}} + \hat{\mathbf{j}})$

(D)	$v(\hat{i} + \hat{j})$
(D)	$\sqrt{2}$

- 18. Two bodies of masses 2 kg and 4 kg are moving with velocities 2 m/s and 10 m/s respectively along same direction. Then the velocity of their centre of mass will be
  - (A) 8.1 m/s
  - 7.3 m/s
  - 6.4 m/s
  - (D) 5.3 m/s
- 19. A stone is just released from the window of a train moving along a horizontal straight track. The stone will hit the ground following
  - (A) Straight path
  - (B) Circular path
  - (C) Parabolic path
  - (D) Hyperbolic path
- 20. A man projects a coin upwards from the gate of a uniformly moving train. The path of coin for the man will be
  - (A) Parabolic
  - (B) Inclined straight line
  - Vertical straight line
  - (D) Horizontal straight line
- 21. A particle (A) is dropped from a height and another particle (B) is thrown in horizontal direction with speed of 5 m/sec from the same height. The correct statement is
  - (A) Both particles will reach at ground simultaneously
  - (B) Both particles will reach at ground with same speed
  - Particle (A) will reach at ground first with respect to particle (C) (B)

- Particle (B) will reach at ground first with respect to particle  $^{(D)}$  (A)
- 22. The angle of projection at which the horizontal range and maximum height of projectile are equal is
  - (A) 45°
  - $\theta = \tan^{-1}(0.25)$
  - $\theta = \tan^{-1}(4)$
  - (D) 60°
- 23. At the top of the trajectory of a projectile, the acceleration is
  - (A) Maximum
  - (B) Minimum
  - (C) Zero
  - (D) §
- 24. An object is projected at an angle of 45 with the horizontal. The horizontal range and the maximum height reached will be in the ratio
  - 1:2
  - (B) 2: 1
  - (C) 1:4
  - (D) 4: 1
- A shell is fired at an angle of 30° to the horizontal with velocity 196 m/s. The time of flight is  $\left[\sin 30^{\circ} = \frac{1}{2} = \cos 60^{\circ}\right]$ 
  - (A) 10 s
  - (B) 16.5 s
  - (C) 20 s
  - (D) 6.5 s
- 26. In which substance does nitrogen exhibit the lowest oxidation state?
  - (A) Nitrogen gas

- (B) Ammonia
- (C) Nitrous oxide
- (D) Nitric oxide
- 27. Among the following, the paramagnetic substance is,
  - (A) Ca<sup>2+</sup>
  - (B) Mg
  - (C) Zn
  - (D) Na
- 28 Which is the most acidic oxide?
  - (A) Cl<sub>2</sub>O
  - (B) Cl<sub>2</sub>O<sub>3</sub>
  - (C) Cl<sub>2</sub>O<sub>5</sub>
  - (D) Cl<sub>2</sub>O<sub>7</sub>
- 29. What is the unit of  $K_p$  for the reaction?

$$CS_2(g) + 4H_2(g) \Longrightarrow CH_4(g) + 2H_2S(g)$$

- (A) Atm
- (B) Atm<sup>-2</sup>
- (C) Atm<sup>2</sup>
- (D) Atm<sup>-1</sup>
- In third row of periodic table the atomic radii from Na to Cl:
  - (A) continuously decreases
  - (B) continuously increases
  - (C) remains constant
  - (D) increases but not continuously
- 31. The elements with atomic numbers 9, 17, 35, 53, 85 are all:
  - (A) Halogens
  - (B) Noble gases
  - (C) Heavy metals
  - (D) Light metals
- 32. For which of the following reactions  $K_P = K_C$

- (A)  $2NOCl(g) \Rightarrow 2NO(g) + Cl_2(g)$
- (B)  $N_2(g) + 3H_2(g) \Rightarrow 2NH_3(g)$
- (C)  $H_2(g) + Cl_2(g) \rightleftharpoons 2HCl(g)$
- (D)  $N_2O_4(g) \rightleftharpoons 2NO_2(g)$
- 33. The oxidation number of oxygen atom in  $O_2^{2-}$  ion is
  - (A) -3
  - (B) -1
  - (C) -5
  - (D) -2
- For the reversible reaction,

$$N_{2(g)} + 3H_{2(g)} \Longrightarrow 2NH_{3(g)} + heat$$

The equilibrium shifts in forward direction

- (A) By increasing the concentration of NH<sub>3(g)</sub>
- (B) By decreasing the pressure
- (c) By decreasing the concentrations of  $N_{2(g)}$  and  $H_{2(g)}$
- (D) By increasing pressure and decreasing temperature.
- In the equilibrium reaction involving the dissociation of  $CaCO_3$ .  $CaCO_3$  (s)  $\rightleftharpoons$  CaO (s) +  $CO_2$  (g)

the equilibrium constant is given by

(A) 
$$\frac{P_{CaO} \times P_{CO_2}}{P_{CaCO}}$$

$$(B)$$
  $C_{CaO} \times \frac{P_{CO_2}}{C_{CaCO_3}}$ 

- (C)  $\frac{P_{CaO}}{P_{CaO}}$
- (D)  $P_{CO_2}$
- 36. At STP, 5.6 L of a gas has a mass of 60 g. The vapour density of the gas is
  - (A) 30

- (B) 60
- (C) 120
- (D) 240
- 37. For the following reaction which is a correct option:-

$$CO_{2(s)} \rightleftharpoons CO_{2(g)}$$

- (A)  $\Delta H > 0$ ,  $\Delta S > 0$  and  $\Delta G < 0$
- (B)  $\Delta H > 0$ ,  $\Delta S > 0$  and  $\Delta G > 0$
- (C)  $\Delta H < 0$ ,  $\Delta S > 0$  and  $\Delta G < 0$
- (D)  $\Delta H > 0$ ,  $\Delta S > 0$  and  $\Delta G = 0$
- Calculate pH of 10<sup>-3</sup> M H<sub>2</sub>SO<sub>4</sub> solution-
  - (A) 3
  - (B)  $3 \log 2$
  - (C)  $3 \log 3$
  - (D) 3.3
- 39. In a reaction of  $H_2O$  (steam) + C (glowing)  $\rightarrow$  CO +  $H_2$ 
  - H<sub>2</sub>O is the reducing agent
  - (B) H<sub>2</sub>O is the oxidising agent
  - Carbon is the oxidising agent
  - Oxidation-reduction does not occur
- 40. Which one of the following is the chiral molecule?
  - CH₃Cl
  - (B) CH<sub>2</sub>Cl<sub>2</sub>
  - CHBr<sub>3</sub>
  - CHClBrI
- 41. 20 cc of CO<sub>2</sub> are passed over red hot coke. The volume of CO evolved is

- (A) 10 cc
- (B) 20 cc
- (C) 30 cc
- (D) 40 cc
- 42. The number of protons, neutrons and electrons in <sup>175</sup><sub>11</sub>Lu, respectively, are:
  - (A) 175, 104 and 71
  - (B) 71, 104 and 71
  - (c) 104, 71 and 71
  - (D) 71, 71 and 104
- The shape of H<sub>3</sub>O+ is
  - (A) tetrahedral
  - (B) pyramidal
  - (C) trigonal planar
  - (D) angular
- 44. Calculate heat of formation of HCl gas from following reaction.  $H_{2(g)} + Cl_{2(g)} \rightarrow 2HCl_{(g)}$ ;  $\Delta H = -194kJ$ 
  - (A) -143 kJ mol<sup>-1</sup>
  - (B) -286 kJ mol<sup>-1</sup>
  - (C) -92 kJ mol<sup>-1</sup>
  - (D) -97 kJ mol<sup>-1</sup>
- 45. Which of the following sets of quantum number represents the highest energy of an atom?
  - (a)  $n = 4, l = 0, m = 0, s = +\frac{1}{2}$
  - (B)  $n = 3, l = 0, m = 0, s = +\frac{1}{2}$
  - (C)  $n = 3, l = 1, m = 1, s = +\frac{1}{2}$
  - (D)  $n = 3, l = 2, m = 1, s = +\frac{1}{2}$
- The electronic configuration of Fe<sup>3+</sup> is
  - (A)  $[Ar]3d^54s^1$

- (B)  $[Ar]3d^64s^0$
- (C)  $[Ar]3d^54s^0$
- (D)  $[Ar]3d^64s^2$
- 47. The wavelength of a spectral line of caesium is 460 nm. What is the frequency of spectral line?
  - $4.5 \times 10^8 \, \text{Hz}$
  - $6.5 \times 10^{14} \text{ Hz}$
  - $3 \times 10^9 \, \text{Hz}$
  - (D) 5.6 × 10<sup>14</sup>
- 48. Which from following molecules has highest bond dissociation enthalph?
  - (A)  $Cl_2$
  - (B) I<sub>2</sub>
  - (C) F<sub>2</sub>
  - (D) Br<sub>2</sub>
- 49. IUPPAC name of the compound



- (A) 2 Bromopentane
- (B) 3 Bromo 2 methylbutane
- (C) 2 Bromo 3 methylbutane
- (D) 1 Bromo 3 methylbutane
- 50. Which among the following molecules is NOT linear in shape?
  - (A) HBr
  - (B) H<sub>2</sub>S
  - (C) BeBr<sub>2</sub>
  - (D) CO<sub>2</sub>

51.		Which of the following is not a true defining property of 'livings'.		(B)	Basal placentation	
				(C)	Axile	
	(A)	Metabolic flux		(D)	Parietal	
	(B)	Cellular organisation	57.		Which of the following statement is incorrect?	
	(C)	Consciousness		(4)	Conidia are produced exogenously and ascospores	
	(D)	Growth		(A)	endogenously.	
52.		Which of the following phase of cell division is longest in oocytes of some vertebrates?		(B)	Yeasts have filamentous bodies with long thread-like hyphae.	
	(A)	Prophase		(C)	Morels and truffles are edible delicacies.	
	(B)	Prophase-I		(D)	Clauiceps is a source of many alkaloids and LSD.	
	(C)	Prophase-II	58.		Find out one (w.r.t. symmetry of flower)	
	(D)	Interkinesis		(A)	Mustard	
53.		Dicot stem is characterised by		(B)	Chilli	
	(A)	The presence of sclerenchymatous hypodermis		(C)	Datura	
	(B)	Semilunar collenchymatous pericycle		(D)	Canna	
	(C)	Parenchymatous pith with no intercellular spaces	59.		Cork cambium is also known as the	
	(D)	Presence of dead cells in pericycle		(A)	Phellogen	
54.		Most extensive metabolic diversity found in:-		(B)	Phelloderm	
	(A)	Protistan		(C)	Phellem	
	(B)	Monerans		(D)	Dermatogen	
	(C)	Mycota member	60.		Each category referred to as unit of classification infact represents a rank and is commonly referred as	
	(D)	Plantae member		(A)	Family	
55.		Phylogeny term represent:-		(B)	Taxon	
	(A)	Life history		(C)	Species	
	(B)	Group of phyla		(D)	Genus	
	(C)	Evolutionary history	61.		Select false Statement:-	
(	(D)	When ovules are borne on central axis and septa is absent, the		(A)	In temperate regions, climatic conditions are not uniform	
56.				(A)	through the year.	
		type of placentation is		(B)	Bark is a non-technical term that refers to all tissues exterior	
	(A)	Free central			to vascular cambium.	

- (C) In the dicot root, vascular cambium ring is completely secondary in origin.
- (D) The spring wood is darker in colour and has low density.
- 62. Read the following statements carefully:-
  - (i) Formation of branch and flower take place by axillary bud
  - (ii) Intercalary meristems are primary meristems because they appear early in life of plant
  - (iii) Intra fascicular cambium is an example of primary meristem
  - (iv) Inter fasicular cambium and cork cambium is known as cylindrical meristem

How many of the above statements are correct?

- (A) One
- (B) Two
- (C) Three
- (D) Four
- Assertion: Ribosomes perform protein synthesis.

Reason: Ribosomes are membrane bound organelle

- (A) Both Assertion and Reason are correct and Reason is the correct explanation of Assertion.
- (B) Both Assertion and Reason are correct but Reason is not the correct explanation of Assertion.
- (C) Assertion is correct but Reason is incorrect.
- (D) Both Assertion and Reason are incorrect.
- Study the organelle given below and identify its function



- (A) It is a site for formation of glycoproteins and glycolipids
- (B) Site for synthesis of steroidal hormone
- (C) These have enzymes that are capable of digesting carbohydrates, proteins, lipids and nucleic acids
- (D) It divides intracellular space into two distinct compartments, i.e., luminal and extra luminal cytoplasm
- Sexual reproduction in fungi follows
  - (A) Plasmo gamy → Meiosis → Karyogamy
  - (B) Plasmo gamy → Karyogamy → Mitosis

- (C) Karyo gamy → Plasmogamy → Meiosis
- (D) Plasmogamy → Karyogamy → Meiosis
- 66. Which two of the below given points are known as the twin characteristics of growth?
  - (i) Increase in mass
  - (ii) Increase in number of individuals
  - (iii) Cellular organization
  - (iv) Cellular differentiation
  - (A) (i) and (ii)
  - (B) (i) and (iii)
  - (C) (ii) and (iii)
  - (D) (iii) and (iv)
- A network of filamentous structure present in the cytoplasm which are involved in many function such as mechanical support, motility, maintains of the shape of the cell. This elaborate network of filamentous are ....... structures present in the cytoplasm is collectively referred to as the ......:
  - (A) Lipoidal, plasma membrane
  - (B) Cytoskeleton, proteinaceous
  - (C) Proteinaceous, centriole
  - (D) Proteinaceous, cytoskeleton
- 68. Which of the following pairs come under the group chrysophytes?
  - (A) Diatoms and Euglena
  - (B) Euglena and Trypanosoma
  - (C) Diatoms and Desmids
  - (D) Gonyaulax and Desmids
- 69. The symbiotic association between fungi and algae is called
  - (A) lichen
  - (B) mycorrhiza
  - (C) rhizome
  - (D) endomycorrhiza
- 70. The primitive prokaryotes responsible for the production of biogas from the dung of ruminant animals, include the

(A)	methanogens		(D)	ligule.
(B)	eubacteria	75.	W	Thich of the following example belong to the same class of
(C)	halophiles	73.		gae?
(D)	thermoacidophiles		(A)	Chara, Fucus, Polysiphonia
W	hich of the following statements is wrong for viroids?		(B)	Volvox, Spirogyra, Chlamydomonas
(A)	They cause infections.		(C)	Porphyra, Ectocarpus, Ulothrix
(B)	Their RNA is of high molecular weight.		(D)	Sargassum, Laminaria, Gracilaria
	They lack a protein coat.	76.		hat is common in all the three, Funaria, Dryopteris and inkgo?
(D)	They are smaller than viruses.		(A)	Independent sporophyte
T	The given figure shows the structure of filamentous blue green		(B)	Presence of archegonia
	lgae. Nostoc with a structure marked as "X". Select the option which shows the correct identification of "X" with its feature.		(C)	Well developed vascular tissues
	3003 <sup>3</sup>		(D)	Independent gametophyte
	×	77.	W	hich one of the following is not a correct statement?
(			(A)	Botanical gardens have collection of living plants for reference.
(A)	Spores - Reproduction		(B)	A museum has collection of photographs of plants and animals.
(B)	Heterocysts - Nitrogen fixation		(C)	Key is a taxonomic aid for identification of specimens.
(C)	Pellicle - Recycling of nutrition		(D)	Herbarium is a store house that contains dried, pressed and preserved plant specimens.
(D)	Mucilaginous sheath - Photosynthesis	78.	W	hich of the following algae contains mannitol as reserve food
Which of the following structure of Prokaryotic cell show resemblance with Mitochondria in function-				aterial ?
16			, ,	Ectocarpus
(A)	Chromatophore		(B)	Gracilaria
(B)	Mesosome		(C)	Volvox
(C)	Ribosome		(D)	Ulothrix
(D)	cell membrane	79.		uring which phase(s) of cell cycle, the amount of DNA in a ell remains at 4C level if the initial amount is denoted as 2C?
Ev	volutionary important character of Selaginella is		(A)	G 0 and G 1
(A)	heterosporous nature		(B)	G 1 and S
(F)	rhizophore		(C)	Only G 2
(B)	•		(D)	G 2 and M
(C)	strobili	80.	In	terkinesis is a

71.

72.

74.

80.

- (A) stage between meiosis I and meiosis II.
- (B) stage between two mitotic divisions.
- (C) interphase.
- (D) both stage between two mitotic divisions and interphase
- 81. Identify the meiotic stage in which the homologous chromosomes separate while the sister chromatids remain associated at their centromeres.
  - (A) Metaphase I
  - (B) Metaphase II
  - (C) Anaphase I
  - Anaphase II
- Match the description (given in column I) with correct stage of prophase I (given column II) and choose the correct option.

#### Column I

#### Column II

- A. Chromosomes are moved
   I. Pachytene to spindle equator
- B. Centromere splits and II. Zygotene chromatids apart
- C. Pairing between homologous III. Anaphase chromosomes takes place
- D. Crossing between homologous IV. Metaphase chromosomes
- (A) A I; B II; C III; D IV
- (B) A II; B III; C IV; D I
- A IV; B III; C II; D I
- A III; B I; C IV; D II

Match the terms (given in column I) with their explanation (given in column II) and choose the correct combination from the options given below.

# Column I Column II (Terms) (Explanation)

- A. Terminalization I. Pairing of homologous chromosomes.
- B. Synapsis II. Point of attachment between homologous chromosomes.
- C. Chiasmata III. Nuclear protein complex that

  helps in adherence of sister

  chromatids and then homologous

  chromosomes.
- D. Synaptonemal IV. Shifting of chiasmata outwards complex towards the ends of a bivalent.
- (A) A IV; B I; C II; D III
- A II; B III; C IV; D I
- A II; B IV; C III; D I
- A IV; B I; C III; D II
- Which one of the following statements is not correct?
  - Each stamen which represents the male reproductive organ consists of a stalk or a filament and an anther.
  - An actinomorphic flower can be dissected into two equal (B) halves from any plane.
  - (C) Superior ovary is found in hypogynous flowers.
  - When stamens are attached to petals, they are epiphyllous as in brinjal.  $^{(D)}$
- 85. Leaves of many grasses are capable of folding and unfolding because they
  - (A) Are very thin
  - (B) Are isobilateral

King locust

Have specialized bulliform cells

91.

Which of the following animal is called a living fossil?

Limulus (B) (D) Have parallel vascular bundles Bombyx (C) When a tree grows older which of the following increase rapidly-Balanoglossus (D) Heart wood (A) Which of the following groups of animals maintain high and Sap wood 92. (B) constant body temperature such as mammals? (C) Pith Reptiles (A) Cortex (D) Amphibians (B) Phenetic classification is based on 87. Birds (C) Observable characteristics of existing entities (D) Fishes The ancestral lineage of existing organisms Respiratory surface for gaseous exchange is decreased in which 93. × (C) disorder: (D) Sexual characteristics Emphysema Regeneration of damaged growing grass following grazing is Asthma (B) largely due to **Bronchitis** (C) Intercalary meristem Pneumonia (D) Lateral meristem The maximum volume of air a person can breathe in after a 94. forced expiration is -Secondary meristem Total lung capacity Apical meristem Inspiration capacity (B) Pick the wrong statement 89 Functional residual capacity (C) Diatoms are primarily produced in the oceans Vital capacity (D) Diatoms are microscopic and float passively in the water 95. Match the column I and Column II and select the correct Walls of diatoms are destructible easily (C) option. Diatomaceous earth is formed by the cell walls of diatoms Column I Column II Euspongia Nematocytes This group is used to represent pathological fungi 90. Pila Flame cells b. Penicillium Planaria (iii) Radula C. Truffles, mushrooms and morels (B) Meandrina d. (iv) Choanocytes Smuts, rusts and moulds a(ii), b(iv), c(i), d(iii) (A) All of the above

(B)

a(iv), b(ii), c(iii), d(i)

	(C)	a(iv), b(ii), c(iii), d(i)	
	(D)	a(iv), b(iii), c(ii), d(i)	
96.		The 'bilateral symmetry' refers	
	(A)	When the body can be divided into two unequal halves on passing central axis through it	
	(B)	To any plane passing through centre, which does not divide the body into equal halves	
	(C)	When the body can be divided into identical left and right halves only in one plane	
	(D)	Any plane passing through the central axis of the body dividing the organism into two equal halves	
97		Which one of the following statements about the given animals is correct?	
	(A)	Roundworms (Aschelminthes) are pseudocoelomates	
	(B)	Molluscs are acoelomates	
	(C)	Insects are enterococlomates enterocoelomates	
	(D)	Flatworms (Platyhelminthes) are coelomates	
98.	Co	ompound squamous epithelium is found in	
	(A)	stomach	
	(B)	intestine	
	(C)	trachea	
	(D)	pharynx	
19.	Ne	ematocyst does not take part in	
	(A)	Food capturing	
	(B)	Locomotion	
	(C)	Offence and defense	
	(D)	Reproduction	
100.		eathing rate in mammals is controlled by a part of the brain	
	(A)	Thalamus	
	(B)	Hypothalamus	
	(C)	Medulla oblongata	
	(D)	Cerebellum	

## **JEE MAINS RESULT 2023**



BHARUCH TOPPER AAYUSH SINGH 99.90%ile



99.76%ile



HRIDYA PANDYA
99.71%ile



79.30%ile



MUKTIK PATEL 99.25%ile



98.97%ile



PARTH BHATT 98.93%ile



98.90%ile



98.82%ile



98.57%ile



HARDEE PAREKH

### **JEE ADVANCED RESULT 2023**



AIR
1955
AYUSH SINGH



AIR
4559
HRIDAY PANDYA



AIR
4828
AYUSH GARG



AIR
2097\*
NEELABH RANA



9411
TANISH



AIR
11177
POONAM PATIL



AIR
12311
HARDEE PAREKH



AIR

15294

MUKTIK



AIR
17193
MAYANK SINGH



AIR

17686

PARTH BHATT

## 10th GSEB RESULT 2023

### **DISHA JADAV**



563/600 99.84%ile

**SHREYANSH** 



534/600 98.66%ile

**DEV JAYSWAL** 



533/600 98.60%ile

### **OUT OF 40 STUDENTS**

99%ILE & ABOVE - 06 STUDENTS 98%ILE & ABOVE - 11 STUDENTS 95%ILE & ABOVE - 32 STUDENTS

**PARV SHAH** 



559/600 99.77%ile



533/600 98.60%ile

**NEEV VITHLANI UMANG MODI** 



554/600 99.63%ile

**NIRVEE PATEL** 



528/600 98.23%ile

546/600 99.33%ile

**RUDRA D PATEL** 



525/600 98.01%ile



545/600



**TEJ GODASARA** 

99.28%ile

99.28%ile



518/600 97.40%ile



509/600 96.53%ile

# 10th CBSE RESULT 2023

### **ABPS TOPPER**



**MITUL CHOUDHARY 97.60**%

### **OUT OF 105 STUDENTS**

90% & ABOVE - 41 STUDENTS





**97.40**%



SOUMIL BISWAS TRISHA VAISHNAV

**97.20**%



RAHEE

95.60%



GREESA VACHHANI MAAHI PATEL

95.40%



95.20%

### **JAY AMBE TOPPER**



SAANVI SWAIN 95.20%



95.00%



95.00%



95.00%



YAKSH PATEL JAYSINH RAHEVAR 94.80%



JEEYA PATEL 94.80%



94.80%